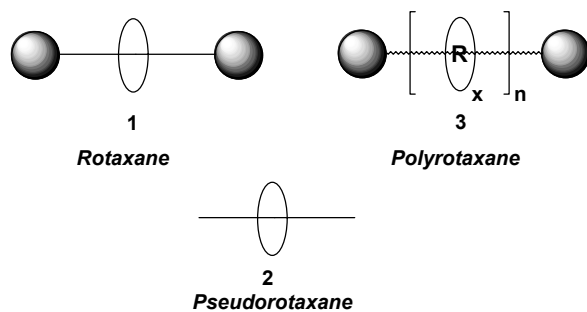


Polyrotaxanes

Polymer Architectures



Methods of preparing mainchain (pseudo)polyrotaxanes

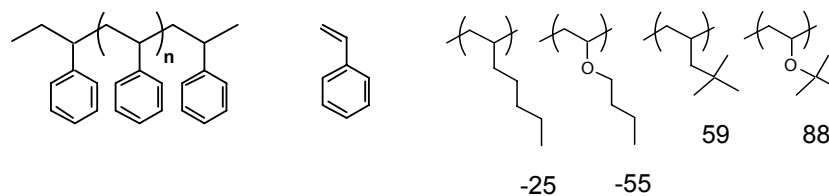
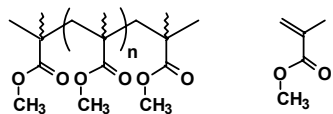
- Cyclisation in the presence of linear macromolecules
- Polymerisation of rotaxanes
- Chemical conversion of rotaxane precursor after polymerisation

Polyrotaxanes: Why?

- Solubilities are altered
- Particular chance of improving solubility and processibility of rigid rod polymers
- Glass transitions follow the Fox equation sometimes
- Glass transition sometimes remain distinct for each species
- Thermal stability can be enhanced
- Control of polymer backbone through cyclics that crystallise
- New types of polymer chain compatibilisation
- "Insulated wire approach"

Consideration

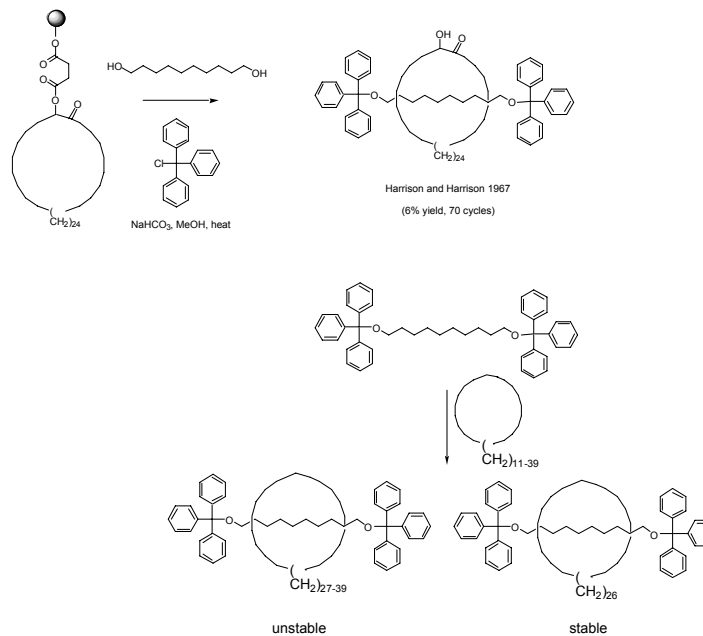
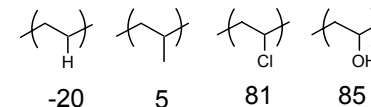
- Blocking groups may be unnecessary due to entrapment of cyclic units in chain folds
- Using rotaxane precursors is unlikely to be of synthetic value

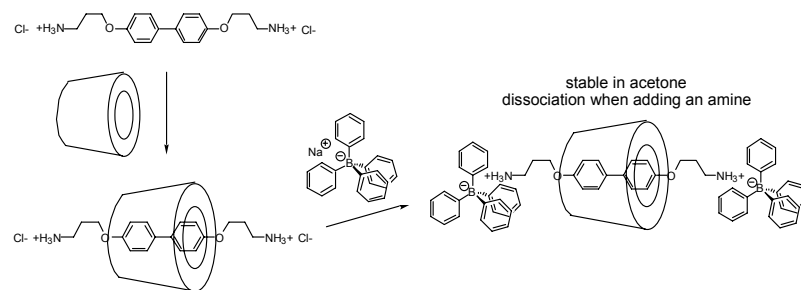
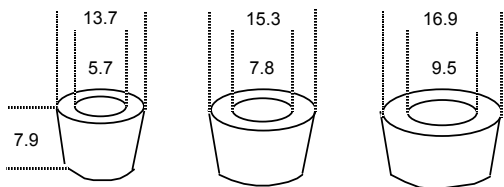
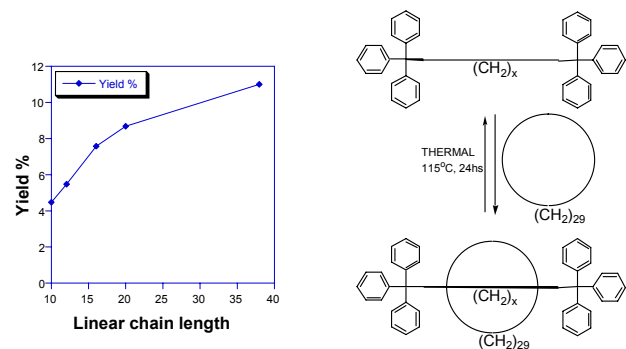
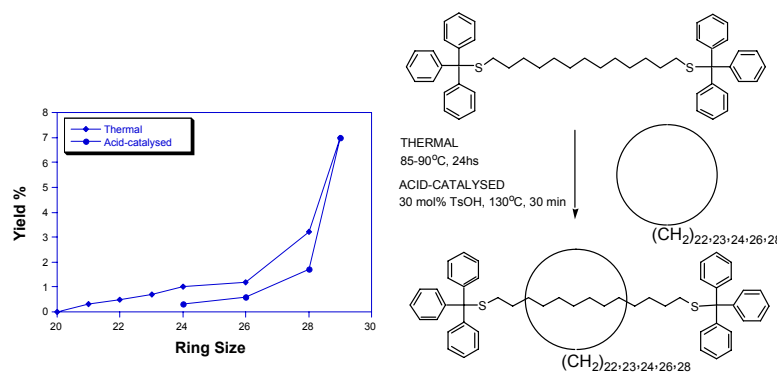


Rotaxanes

In 1910 evoked

In 1967 proposed and proven





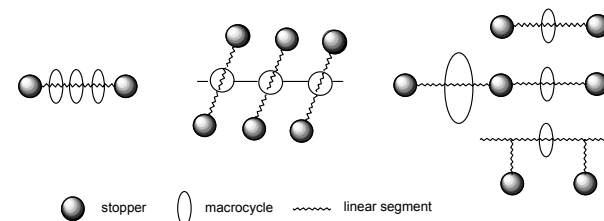
Homorotaxanes and Heterorotaxanes

Homorotaxanes are composed of cyclic and linear species that are chemically equivalent, e.g., a polyrotaxane constructed from poly(ethyleneoxide) and a crown ether^{1,10-11}.

Heterorotaxanes on the other hand, involve cyclic and linear species having different chemical structures, e.g., a polyrotaxane derived from poly(ethylene) and α -cyclodextrin²⁵, affording alternatives to classical copolymer architectures.

Mainchain and Sidechain Polyrotaxanes

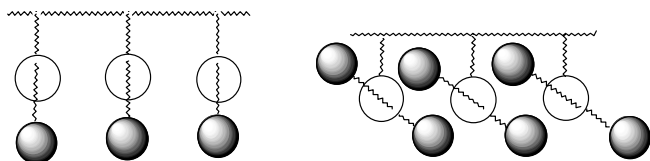
Mainchain Polyrotaxanes



Approaches to mainchain polyrotaxanes

- Polymerisation of preformed rotaxane
- Threading of linear molecule through preformed macrocycle
- Polymerisation of macrocycle in presence of linear species followed by end-capping

Sidechain Polyrotaxanes



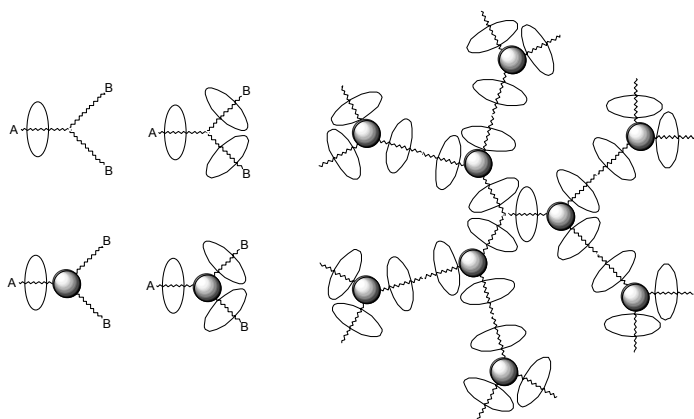
Approaches to sidechain polyrotaxanes - cyclic orthogonal to backbone

- Threading of preformed graft copolymer plus end-capping
- Grafting in presence of cyclic unit and linear link
- Grafting of preformed rotaxane monomer
- Polymerisation of macromonomer in the presence of cyclic unit
- Polymerisation of preformed macromonomeric rotaxane

Approaches to sidechain polyrotaxanes - cyclic parallel to backbone

- Threading of preformed pendant cyclic graft copolymer
- Grafting of cyclic unit in presence of linear species followed by end-capping
- Grafting of preformed rotaxane to linear backbone
- Polymerisation of a preformed species followed by postpolymerisation end-capping
- Polymerisation of preformed rotaxane

Dendritic and Hyperbranched Polyrotaxanes

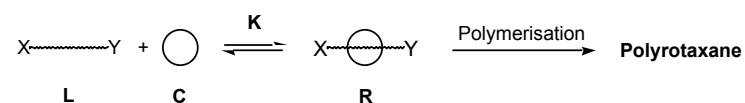


Synthesis of Polyrotaxanes

Statistical Approach

Le Chatelier

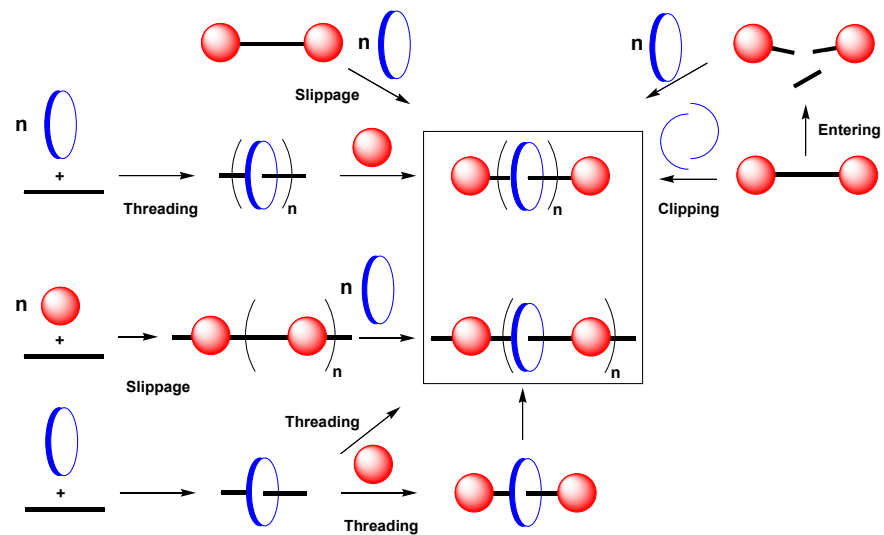
Equilibrium constant (0.05 to 5 l/mol)

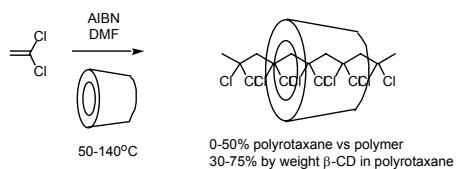


$$K = e^{-(\Delta G/RT)} = [R]/[L][C]$$

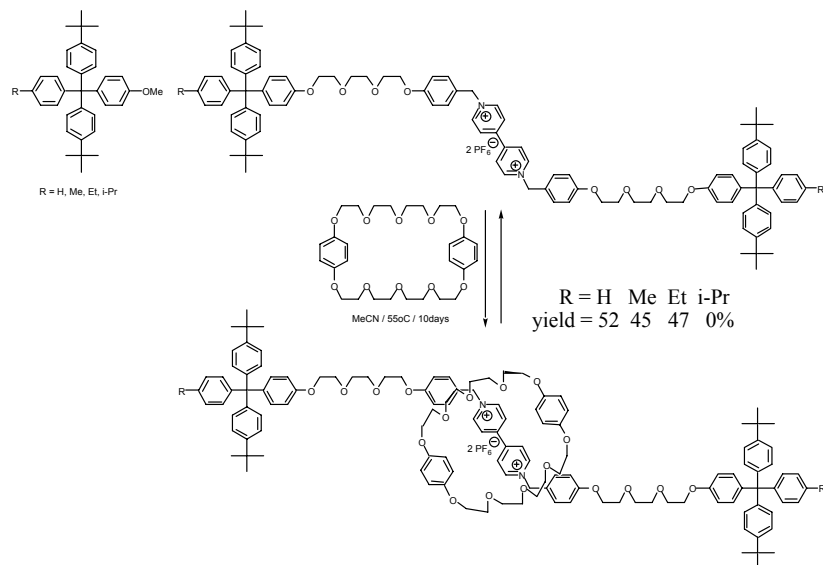
$$\ln K = -(\Delta G/RT) = -(\Delta H - T\Delta S)/RT = -\Delta H/RT + \Delta S/R$$

entropically driven = (ΔH small)



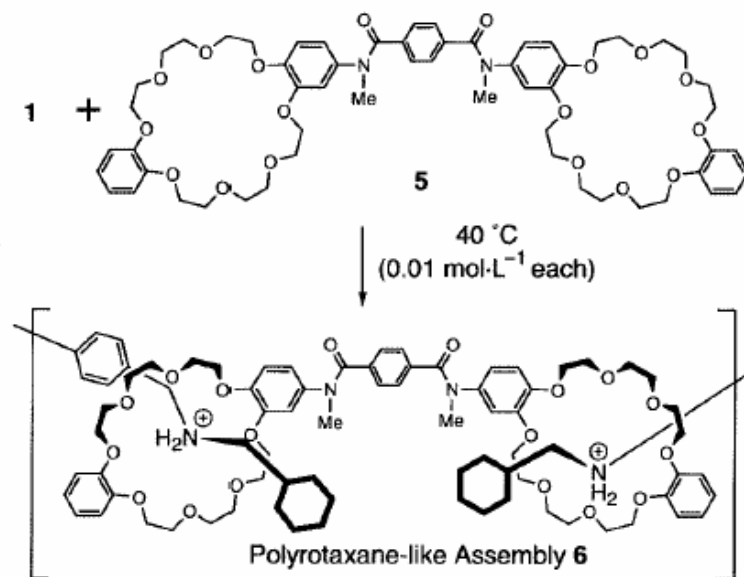
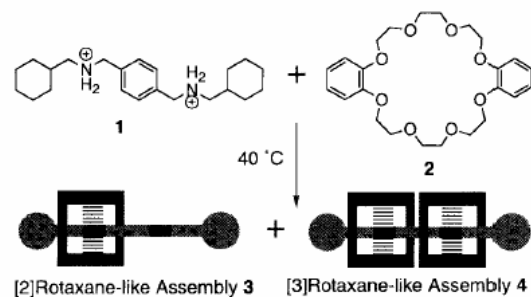


Directed Synthesis
Preformed Monomer
Slippage

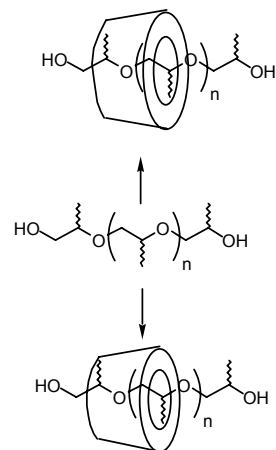
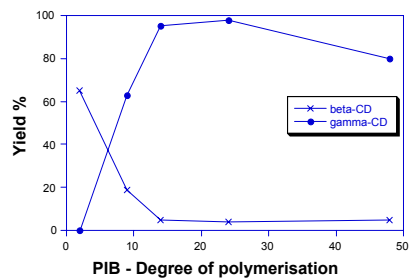
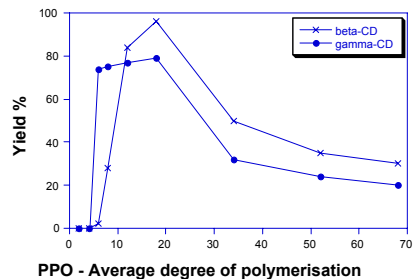


	α -cyclodextrin	β -cyclodextrin	γ -cyclodextrins
PPO (MW 1000)	0%	96%	80%
		2:1 complex	
PEO	~100%	low	~100%
	2:1 complex		2:1 complex (double)

Similarly polyethylene glycols form crystalline complexes with α -CD but not β -CD²¹.



Solvent	Time / d	$M_n (M_w/M_n)$	Higher	Lower
			$M_n (M_w/M_n)$	$M_n (M_w/M_n)$
CD ₃ COCD ₃	113	1500 (1.4)	3300 (1.2)	1200 (1.1)
CD ₃ CN	107	1700 (1.5)	3800 (1.2)	1200 (1.1)

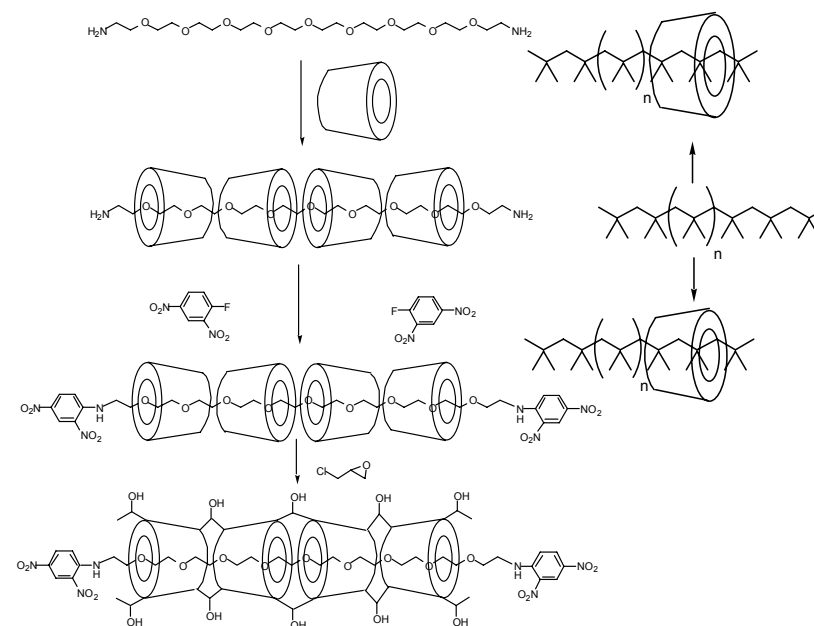


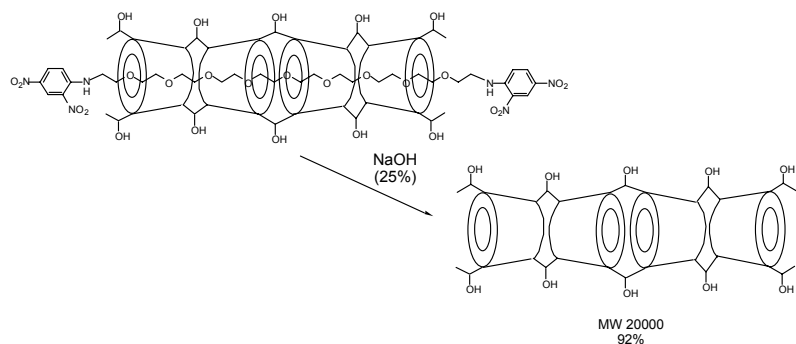
Cyclodextrins and Hydrophobic Polymers

PIB was added to saturated aqueous solutions of both β - and γ -CDs under sonication, the solution became turbid and the complexes were formed as crystalline precipitates.

Polymer/Oligomer	Structure	MW	α -CD	β -CD	γ -CD
			4.5 Å	7.0 Å	8.5 Å
OE	$-\text{CH}_2\text{CH}_2-$	563	63	0	0
Squalane	$-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2-$ CH_2	423	0	62	24
PIB	$-\text{CH}_2\text{CH}_2\text{CH}_2-$ CH_2	~800	0	8	90

the relative sizes of the cavities of cyclodextrins and the cross-sectional areas of the polymers are essential in complex formation in the absence of non-hydrophobic effects.

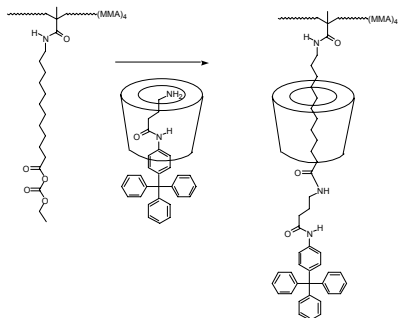




PEG 1450 (approx. 33 repeat units)

about 16 aCDs times MW 1200 = MW 20000

not soluble in water anymore!



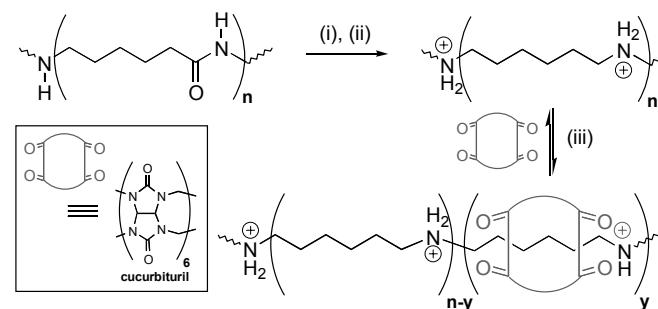
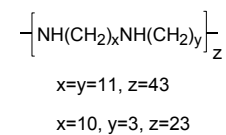
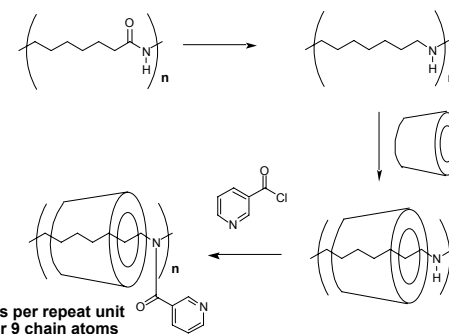
The threading of α -CD was followed by $^1\text{H-NMR}$ spectroscopy, in which significant downfield shifts of CD protons are observed as the polyrotaxane forms. The larger β -CD derivative is threaded several orders of magnitude more rapidly than the smaller α -CD.

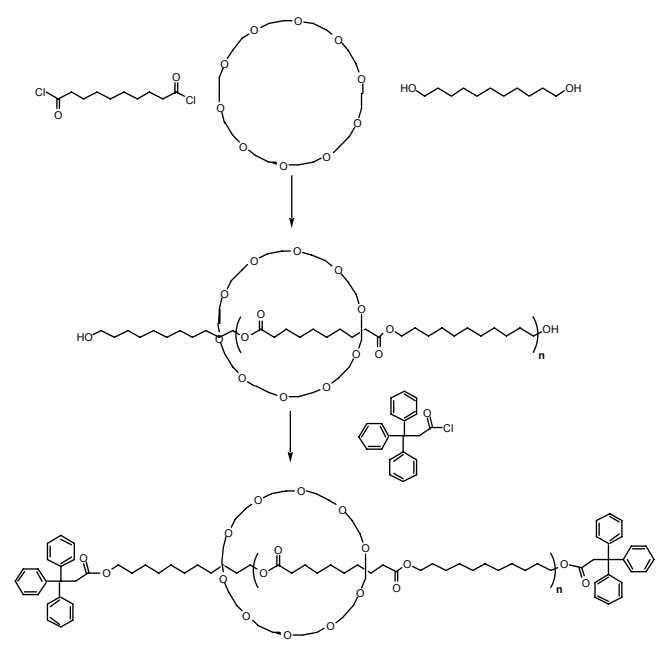
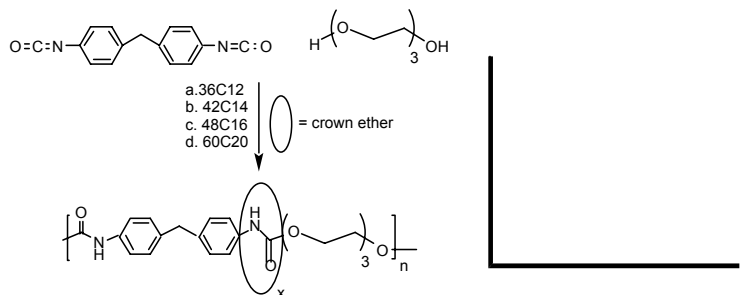
The structure and the molecular weight of the polyamines dramatically affect both threading and dethreading rate (**11** more than two orders of magnitude faster than **12**). Reaction of the amino groups with nicotiny chloride gave stable polyrotaxanes with up to 0.67 α -CD units per amino group in the case of **11**; this corresponds to 37 α -CDs per macromolecule or one α -CD per 9 backbone atoms.

100-fold difference

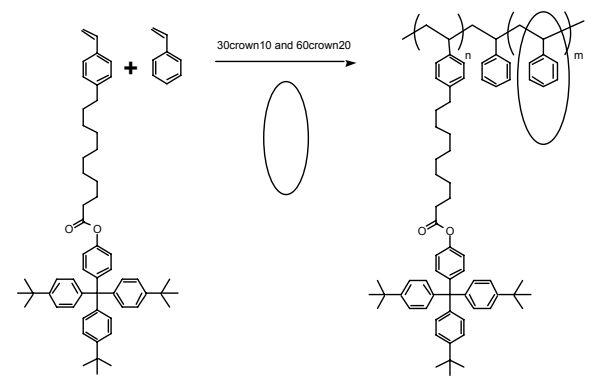
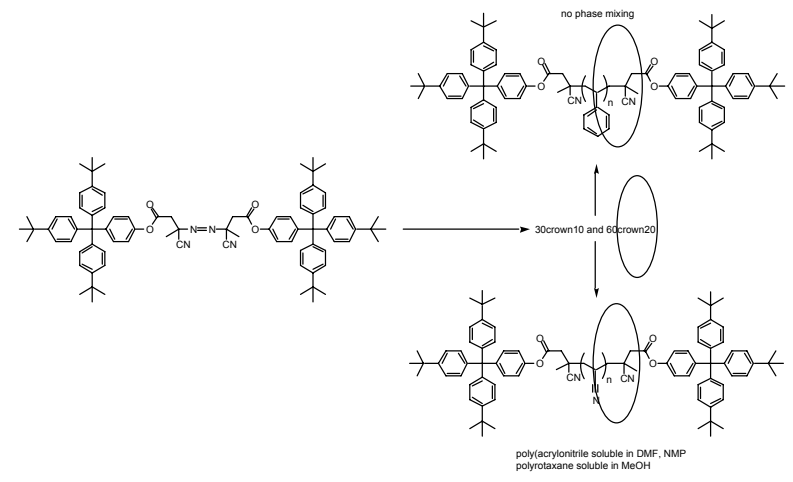
Faster

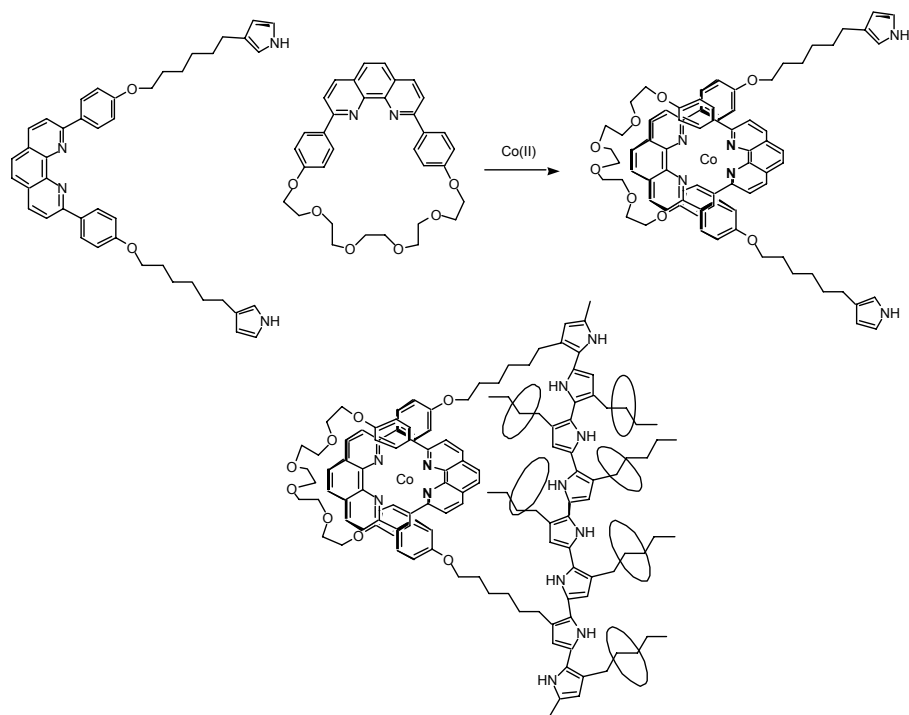
Slower





Plasticisation





TI: ROTAXANES, CATENANES, POLYROTAXANES, POLYCATENANES AND RELATED

MATERIALS

AU: GIBSON_HW, BHEDA_MC, ENGEN_PT

JN: PROGRESS IN POLYMER SCIENCE, 1994, Vol.19, No.5, pp.843-945

TI: alpha Cyclodextrin/polyethylene glycolpolyrotaxane: A study of the threading process

AU: Ceccato_M, LoNostro_P, Baglioni_P

JN: LANGMUIR, 1997, Vol.13, No.9, pp.2436-2439

TI: Self-threading-base approach for branched and/or cross-linked poly(methacrylate rotaxane)s

AU: Gong_CG, Gibson_HW

JN: JOURNAL OF THE AMERICAN CHEMICAL SOCIETY, 1997, Vol.119, No.25,

TI: Design and construction of supramolecular architectures consisting of cyclodextrins and polymers

AU: Harada_A

JN: ADVANCES IN POLYMER SCIENCE, 1997, Vol.133, pp.141-191

TI: Photochemical synthesis of polyrotaxanes from stilbene polymers and cyclodextrins

AU: Herrmann_W, Schneider_M, Wenz_G

JN: ANGEWANDTE CHEMIE-INTERNATIONAL EDITION IN ENGLISH, 1997, Vol.36, No.22, pp.2511-2514

Interlocked macromolecules

Raymo, F. M., Stoddart, J. F Chem. Rev 1999, **99**, 1643

Encapsulated conducting polymers

Cardin, D. J., Adv. Mater. 2002, **14**, 553